

K. N. Toosi University of Technology Faculty of Materials Science and Engineering



Materials Characterization Methods

Fifth Session (Application of X-ray in Materials Analysis)

Reza Eslami-Farsani



Quantitative Analyses of Materials Using X-Ray Diffraction



- Using the diagram of radiation intensity in terms of 2θ from the diffractometry method or the Debby Scherrer method, the percentage of different materials in the study composition can be calculated with XRD.
- For each test sample, it is necessary to prepare standard samples whose composition is the same as the composition of the test sample but with different percentages.





• For a combination of A, B and C, if we want to find the percentages of components, we need to prepare at least 4 or 5 combinations with specific percentages of A, B and C as standard compounds. Then the standard samples are irradiated with X-rays and the curves are compared with the curves of the original sample and the percentage of compounds in the sample is determined by means of related calculations.







- Bragg's law assumes that the conditions of constructive interference are met, namely: The crystal is perfect and there are no defects in it. The beam is monochromatic and strikes the sample in parallel.
- In practice, these conditions never exist completely.



Company Logo



Destructive Interference

- X-ray diffraction under non-ideal conditions (angles close to Bragg's angle). The input rays are not parallel. If the phase difference between the two waves is large, then we will not have diffraction.
- In the case of destructive interference, if the phase difference is very small, we will have diffraction, but the peak will wide. This means that fewer waves are eliminated and more angles (close to 2θ) are diffracted with less intensity (due to non-compliance with Bragg's law).





- \checkmark The peak at certain angles represents the crystalline structure.
- Amorphous and liquids materials both have structures characterized by a lack of almost complete periodic order.
- There is no periodic structure in monoatomic gases. In such gases, the atoms are randomly placed together. Their scattering curve not only does not show a pick, but also shows a regular decrease in intensity with increasing scattering angle.







Peak Broadening =

Used Equipment + Decrease in Crystallite Size + Lattice Strain

 $\beta_{\text{observed}} = \beta_{\text{instrumental}} + \beta_{\text{strain and particle size}}$

 $\beta_r = \beta_{\text{crys. size}} + \beta_{\text{strain}}$

Company Logo



Other Relationships:

$$\beta_r^2 = \beta_o^2 - \beta_i^2$$

$$\beta_r = \sqrt{(\beta_\circ - \beta_i \sqrt{(\beta_\circ^2 - \beta_i^2)})}$$

Company Logo



Scherrer Formula: Peak broadening equation only due to crystallite size.



Best Case: Grain size range between 2 to 300 nm



If peak broadening is due to the strain:

$$\beta_{strain} = \eta \tan \theta$$

η= Strain in material

Deformation causes a local change in atomic spaces, and these

lattice strains cause the peak broadening.







$$\beta_r = \beta_{\text{crys. size}} + \beta_{\text{strain}}$$

$$\beta_r = \frac{k\lambda}{D\cos\theta} + \eta \tan\theta \implies \beta_r \cos\theta = \frac{k\lambda}{D} + \eta \sin\theta$$





Example:







Annealed particles	I		CuKa radiation:	1.54056 Å
Peak#	20 (°)	hki	FWHM (*)	FWHM (radians) = β_{f}
1	64.9	200	0.5	8.73E-03
2	82.0	21 1	0.6	1.05E-02
3	98.5	220	1.1	1.92E-02

Deformed

particles

CuKa 1.54056 Å

Peak #	2 0 (°)	sin O	hkJ	FWHM (°) = β_{o}	FWHM (radians) = β_{o}	$\boldsymbol{\beta}_r = \sqrt{\boldsymbol{\beta}_s^2 - \boldsymbol{\beta}_l^2}$	β,cosθ
1	64.9	D.5366	200	0.75	1.31E-02	9.76E-03	8.23E-03
2	82.D	D.6561	21 1	1.0	1.75E-02	1.40E-02	1.05E-02
3	98.5	D.7576	220	1.5	2.62E-02	1.78E-02	1.16E-02



Company Logo

19

